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A neutralization reaction occurs between an acid and a base and produces water and a salt. The final solution is closer to a neutral pH than either the acid or base alone. Hydrochloric acid (HCI) reacts with potassium hydroxide (KOH), producing a solution of potassium chloride (KCI) and water (H. 2O).

Answers Acids, Bases, and Solutions

"Acids, Bases and Salts" focuses on acids and bases, and Salts in solutions. These NCERT Solutions comprises answers to all the questions of the chapter that are there in the NCERT textbook.

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