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(Hobbes Leviathan,
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Chapter 13 States Of
Matter

Start studying States
of Matter (chapter
13). Learn
vocabulary, terms,
and more with
flashcards, games,
and other study tools.

States of Matter

Page 12/33

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Chapter 13

(Chapter 13) Matter

Flashcards | Quizlet

You are already

familiar with the

three common states

of matter: solid,

liquid, and gas. Solid

objects litter the room

around you. For

example, you can

easily recognize the

shape of your desk;

you know that your

backpack cannot hold

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States of Matter
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seven textbooks. You encounter liquids throughout the day as you

Chapter 13: States of Matter

Chapter 13 States of Matter
137 SECTION
13.1 THE NATURE OF GASES (pages 385–389) This section introduces the kinetic theory and

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describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385–386) 1.

Name Date Class

STATES OF MATTER

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13 States Of Matter

There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can refer you to somebody.) Those three states are solid, liquid, and gas.

These three states are quite different. The main difference is in

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States of Matter
their particles.

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Chapter 13: States of
Matter - Chemistry by
Anna

Chapter 13 States Of
Matter Chapter 13
States of Matter137

SECTION 13.1 THE
NATURE OF GASES

(pages 385–389)

This section

introduces the kinetic
theory and describes

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how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance.

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Matter Chapter 13
States of Matter 137

SECTION 13.1 THE
NATURE OF GASES
(pages 385–389)

This section
introduces the kinetic
theory and describes
how it applies to
gases. It defines gas
pressure and explains
how temperature is
related to the kinetic

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energy of the particles of a substance.

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Chapter 13 States Of Matter Worksheet

Chapter 13: States of Matter. STUDY. PLAY.

Kinetic Molecular

Theory. Explains the properties of gases in terms of the energy, size, and motion of their particles. Elastic

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Collision. Describes a collision in which kinetic energy may be transferred between colliding particles but the total kinetic energy of the two particles remains the same.

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Chemistry (12th

Page 22/33

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Chapter 13

Edition) answers to
Chapter 13 - States of
Matter - 13.1 The
Nature of Gases -
13.1 Lesson Check -
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ISBN-13:
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Chemistry (12th
Edition) Chapter 13 -
States of Matter ...

all matter consists of
tiny particles that are
constantly in motion

What are the three
assumptions of the
kinetic theory as it
applies to gases? -The
particles in a gas are

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States of Matter

Considered to be small, hard spheres with an insignificant volume. -The motion of the particles in a gas are rapid, constant, and random.

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The Sustainable Development Goals

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are a call for action by all countries – poor, rich and middle-income – to promote prosperity while protecting the planet. They recognize that ending poverty ...

United Nations

Sustainable

Development – 17

Goals to ...

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Matter. 13.1 The

Nature of Gases -

Chemistry & You;

13.1 The Nature of

Gases - Sample

Problem 13.1; 13.1

The Nature of Gases -

13.1 Lesson Check;

13.2 The Nature of

Liquids - Chemistry &

You; 13.2 The Nature

of Liquids - 13.2

Lesson Check; 13.3

The Nature of Solids -

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Chemistry & You;

13.3 The Nature of
Solids - 13.3 Lesson

Check; 13.4 Changes
of State - Chemistry &
You

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Matter. 13.1 The
Nature of Gases -
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13.1 The Nature of

Gases - Sample

Problem 13.1; 13.1

The Nature of Gases -

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You; 13.2 The Nature

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States of Matter 1

Chapter 13 States of
Matter 2 Kinetic

Theory as Applied to

Gases Fundamental
assumptions about

gases. The particles in
a gas are considered

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Chapter 13

to be small, hard
spheres with an
insignificant volume.

Between particles in a
gas there is empty
space. No attractive
or repulsive forces
exist between the
particles. 3

Chapter 13 States Of
Matter Worksheet

The attacks the USPS
continues to face are

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not just attacks on the postal service but attacks on Black lives. To defund the USPS would be to deny future generations this opportunity and dishonor the legacy of Black postal workers. Now, we 're taking this matter into our own hands by writing and sending

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#BlackLoveLetters
through USPS...

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