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the center of the atom, the nucleus, with empty space around it.

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model of the atom: Protons and neutrons go in the nucleus (center). Electrons go in energy levels of the electron cloud.

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Answer Key A. Structure of an atom 1.
Fill in the blanks concerning sub-atomic particles. (a) There are three sub-atomic particles in atoms. They are _ protons _, _ neutrons _ and _ electrons _.

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tells you the number of _____ in a neutral atom of that element. The atomic number gives the “identity” of an element as well

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as its location on the periodic table. No two different elements will have the _____ atomic number. 4. The _____ of an element is the average mass of an element's naturally occurring atom, or

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~~Answer Key~~
Name three sub atomic particles of an atom. 2. Define atomic number and mass number. 3. Describe the particle scattering experiment conducted by Rutherford with a diagram. 4.

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(1) Write electronic configuration of element X which has atomic number = 15. Write its valency also.

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General Chemistry Mr.

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Worksheet 1. The number of protons in the nucleus of an atom of a particular element is called the _____ of that element. 2.

In a neutral element, the number

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of protons must equal the number
of _____. 3.

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