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When you put a little amount of blue copper sulphate in a clean test tube, then heat the content of the test by using a flame, A black substance will be formed, Blue

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copper sulphate decomposes by the heat into copper oxide

(black colour)

and sulphur trioxide. CuSO_4

$\text{CuO} + \text{SO}_3 \text{H}_2\text{O} +$

$\text{SO}_3 \rightarrow \text{H}_2\text{SO}_4$

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